Effects of Nitrogen contamination in liquid Argon

WArP Collaboration

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Abstract

A dedicated test of the effects of Nitrogen contamination in liquid Argon has been performed at the INFN-Gran Sasso Laboratory (LNGS, Italy) within the WArP R&D program.

A detector has been designed and assembled for this specific task and connected to a system for the injection of controlled amounts of gaseous Nitrogen into the liquid Argon. Purpose of the test is to detect the reduction of the Ar scintillation light emission as a function of the amount of the Nitrogen contaminant injected in the Argon volume. A wide concentration range, spanning from $\sim 10^{-1}$ ppm up to $\sim 10^3$ ppm, has been explored.

Measurements have been done with electrons in the energy range of minimum ionizing particles (γ -conversion from radioactive sources). Source spectra at different Nitrogen contaminations are analyzed, showing sensitive reduction of the scintillation yield at increasing concentrations.

The rate constant of the light quenching process induced by Nitrogen in liquid Ar has been found to be $k(N_2)=0.11 \ \mu s^{-1} ppm^{-1}$.

Direct PMT signals acquisition at high time resolution by fast Waveform recording allowed to extract with high precision the main characteristics of the scintillation light emission in pure and contaminated LAr. In particular, the decreasing behavior in lifetime and relative amplitude of the slow component is found to be appreciable from $\mathcal{O}(1 \text{ ppm})$ of Nitrogen concentrations.

Key words: Scintillation Detectors, Liquid Noble Gases, Quenching (fluorescence), Dark Matter Search

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1 Introduction

Dark Matter search in the form of WIMPs is of primary interest in the present astroparticle physics scenario. Direct detection of Dark Matter with noble gases liquified as target medium is one of the most promising line of

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development in experimental technology. Argon is an ideal medium and the feasibility of Ar-based detectors has been firmly proved by the R&D study of the WArP Collaboration [1]. The two-phase (liquid-gas) technology developed by WArP is based on the simultaneous detection of both signals produced by ionization events in liquid Argon (LAr): free electron charge and scintillation light.

In the case of the ionization charge, the main limitation to its full collection comes from electron attachment process due to contaminations of electronegative impurities like Oxygen present at residual level in (commercial) Argon. Detailed studies have been performed and led, since the pioneering work of the Icarus Collaboration [2], to the development of adequate purification systems (Oxygen reactants and molecular sieves, down to ≤ 0.1 ppb of Oxygen equivalent concentration), normally employed in experimental applications.

On the other hand, the effects of impurities on the scintillation light yield are rather less precisely explored. Quenching (i.e. non-radiative) process in two-body collision of impurity-molecules with Ar₂ excimer states, otherwise decaying with VUV light emission, may take place depending on the type of impurity and the concentration level. This led to include within the WArP R&D program investigations on Oxygen and Nitrogen contamination effects on LAr scintillation light. Two experimental tests have been thus performed.

The case of Nitrogen (N_2), present in commercial Argon at the level of 1 to 10 ppm, depending on the selection grade, is the subject of the present study (while results from the test with Oxygen have been reported separately [3]). Quenching of the light yield in N_2 -contaminated LAr has been found, in particular of the slow component of the scintillation emission. This effect is relevant not only because it imposes some limitations on the full collection of the light available, but also because it deteriorates the detector capability of Pulse Shape discrimination of background events (electrons from γ interactions) from the signal (Ar-recoils, potentially induced by WIMP interactions) [1].

2 Features of the scintillation radiation in LAr

It is experimentally established and theoretically well understood that recombination and de–excitation processes following the passage of ionizing particles in liquid Argon (LAr) finally originate scintillation radiation [4], [5]. In particular, interactions of ionizing particles in LAr cause the formation of both electron-hole (Ar⁺) pairs and Ar* excited atoms (the ratio Ar*/Ar⁺ being 0.21 [6]). Ar* excited atoms lead to the formation of the Ar₂* low excited dimer through collision with Ar atoms ("self-trapping" process). On the other hand,

 ${\rm Ar}^+$ ions also lead to the formation of ${\rm Ar}_2^*$ through a number of subsequent processes including recombination [7]. The excited dimer states formed in both cases in LAr are recognized to be the singlet ${}^1\Sigma_u$ and the triplet ${}^3\Sigma_u$ excimer states in the M-band, characteristic of the solid Argon structure [8]. The rise-time corresponding to excimer formation and relaxation is very fast, in the sub-nanosecond range for both components.

The de-excitation processes to the dissociative ground state ${}^{1}\Sigma_{g}$ lead to scintillation light emission in the Vacuum Ultra-Violet (VUV) region:

$$Ar_2^* \rightarrow 2 Ar + 1 \gamma$$
 (1)

In liquid phase, various electron-to-ion recombination mechanisms along the ionization track are active, depending on the type of the ionizing particle and on its LET (Linear Energy Transfer, i.e. the specific energy loss along the path) [6]. These mechanisms affect the number of excited dimers Ar_2^* produced per unit of deposited energy, as well as the relative populations of the singlet and triplet states. For a minimum ionizing particle (mip) the photon yield has been measured to be $4.0 \times 10^4 \ \gamma/\text{MeV}$ [6].

The γ decay spectrum of both excimer states has been extensively investigated. It is well represented by a gaussian shape, peaking at $\lambda=127$ nm with $FWHM\simeq 6$ nm (at boiling point T=87.3 K, $\rho=1.395$ g/cm³) [9]. The time dependence of the photon emission is known less precisely. The various measured values of both the excimer lifetimes and of the relative amplitudes reported in literature are in fact quite different with each other. To a first approximation, in all measurements the overall scintillation light emission exhibits a double exponential decay form (see [9] for a complete compilation of available data), characterized by two very different components: a fast component, with a time constant τ_S in the 2 ns to 6 ns range, and a slow component, with a time constant τ_T in the 1100 ns to 1600 ns range. These are associated to the lifetimes of the singlet $^1\Sigma_u$ and of the triplet $^3\Sigma_u$ states in LAr, respectively.

In addition to these fast and slow components, an *intermediate* component (with decay time around 40 ns) has been also sometime reported in literature [11] [14], the origin of which was never investigated in details nor definitively confirmed.

While the time constants do not depend appreciably on the ionization density, the amplitude ratio A_S/A_T of the singlet and triplet states is found to be strongly affected by the ionization density [10]. In particular, all measurements show an enhancement of the $^1\Sigma_u$ formation (fast component) at higher deposited energy density 1 . As an example, the relative amplitude for the fast and for the slow component in case of a mip is reported to be $A_S/A_T = 0.3$ [11] (i.e. $A_S = 23\%$ and $A_T = 77\%$ respectively), while for heavily ionizing

This effect is contrary to the effect in organic scintillators, in which the relative intensity of the slow component increases with increasing specific ionization density.

particles the intensity ratio increases (e.g. $A_S/A_T = 1.3$ for α -particles and =3 for nuclear recoils, but higher values are given elsewhere [12]). This wide separation is an important feature of the scintillation signals in LAr, leading to define robust Pulse Shape Discrimination criteria suitable for particle identification.

The lack of accuracy/consistency in the presently available data can presumably be ascribed to (i) light quenching from residual concentration of impurities diluted in LAr, (ii) limited time resolution employed in past experiments, (iii) dependence on the LAr density (not systematically reported or accounted for).

2.1 Light quenching from diluted N_2 impurities in LAr

The de-excitation process rate can be described by a first-order rate law, characterized by two decay time constants τ_S and τ_T associated to the singlet ${}^{1}\Sigma_u$ and the triplet ${}^{3}\Sigma_u$ excimer states of Ar₂*, respectively (j = S, T):

$$\frac{d \left[Ar_2^*\right]_j}{dt} = -\frac{1}{\tau_i} \left[Ar_2^*\right]_j \quad \Rightarrow \quad [Ar_2^*]_j(t) = \left[Ar_2^*\right]_j(0) e^{-t/\tau_j} \tag{2}$$

Assuming that one scintillation photon per Ar_2^* dissociation is produced, the time dependence of the scintillation light emission in *pure* LAr can thus be represented by the following probability distribution function (pdf):

$$\ell(t) = \frac{A_S}{\tau_S} \exp(-\frac{t}{\tau_S}) + \frac{A_T}{\tau_T} \exp(-\frac{t}{\tau_T})$$
 (3)

with the sum of the relative amplitudes constrained to unity $(A_S + A_T = 1)$ so that

$$\int_0^\infty \ell(t)dt = (A_S + A_T) = 1. \tag{4}$$

Light quenching due to (residual and unknown) impurity concentration in LAr is often indicated as the dominant source of uncertainty in the available data on LAr intrinsic properties.

Residual concentration at the ppm (part per million) level of N₂, O₂, H₂O and CO+CO₂ contaminants (whose presence is usually indicated in commercially available Argon) can lead to a substantial reduction of the scintillation light intensity, mainly due to the decrease of the *slow* component amplitude.

Argon purification systems (Oxygen reactants and molecular sieves) are known to be very effective in reducing the O_2 and H_2O , $CO+CO_2$ contamination up to a negligible level (≤ 0.1 ppb, part per billion). At present, methods for removing the N_2 residual content in Argon are instead less commonly used

in experimental applications 2 , and the effects of N_2 contamination are rather poorly known.

The main reaction to be considered is the quenching process in two-body collision of N_2 impurities with Ar_2^* excimer states [14]:

$$Ar_2^* + N_2 \rightarrow 2Ar + N_2$$
 (5)

This non-radiative collisional reaction is in competition with the deexcitation process leading to VUV light emission. As a result, a sensitive quenching of the scintillation light yield is expected, in particular for lightly ionizing particles. In fact, the quenching process leads to the decrease of the excimer concentration $[Ar_2^*]$, while the contaminant concentration $[N_2]$ stays constant in time. To a first approximation, also for this case a first-order rate law can be assumed and characterized by the rate constant k:

$$\frac{d \left[Ar_2^* \right]}{dt} = -k \left[N_2 \right] \left[Ar_2^* \right] \quad \Rightarrow \quad \left[Ar_2^* \right](t) = \left[Ar_2^* \right](0) e^{-k[N_2] t} \quad (6)$$

The value reported in literature is $k \simeq 3.8 \times 10^{-12} \text{ cm}^3 \text{ s}^{-1}$ [14] (equivalent to 0.11 ppm⁻¹ μs^{-1}).

The time dependence of the scintillation light emission in N_2 -contaminated LAr can be represented by an expression formally identical to Eq.3:

$$\ell'(t) = \frac{A_S'}{\tau_S'} \exp(-\frac{t}{\tau_S'}) + \frac{A_T'}{\tau_T'} \exp(-\frac{t}{\tau_T'})$$
 (7)

where τ'_j [j = S, T], the *effective* (decreased) lifetime of each component, is a function of the [N₂] concentration:

$$\frac{1}{\tau_j'}([N_2]) = \frac{1}{\tau_j} + k [N_2] \tag{8}$$

and A'_{j} , the *effective* (quenched) amplitude, is also function of the $[N_{2}]$ concentration:

$$A'_{j}([N_{2}]) = \frac{A_{j}}{1 + \tau_{j} \ k \ [N_{2}]}$$
 (9)

Notice that under these definitions $\ell'(t)$ reduces exactly to $\ell(t)$ for $[N_2]=0$, namely for pure Argon. The sum of the quenched amplitudes of Eq. (7) is no longer constrained to unity:

$$\int_0^\infty \ell'(t)dt = (A_S' + A_T') \le 1 \tag{10}$$

 $^{^2}$ O₂ and CO+CO₂ are known to be electro-negative molecules. Their high e-affinity induces ionization electron attachment, with dramatic effects on detectors based on charge collection (LAr-TPC). N₂ molecules are not electro-negative [15], and therefore no appreciable effects on the free electron charge can be expected.

From Eq. (10) it can be defined the overall quenching factor (Q_F) :

$$Q_F = A_S' + A_T' \; ; \quad 0 \le Q_F \le 1$$
 (11)

namely, the ratio between the total intensity of scintillation light emitted for a given $[N_2]$ contamination with respect to the case of pure liquid Argon. Q_F also represents by definition the normalization value for $\ell'(t)$.

This work is dedicated to a systematic study of the the effects of Nitrogen contamination in LAr. In particular, the aim is:

- to determine the overall quenching factor affecting the LAr scintillation light yield at various N₂ concentrations [Sect. 4];
- to determine the effective lifetimes τ'_j as a function of the N₂ contamination for each component, as well as the quenched amplitudes A'_j [Sect. 5];
- to extract the main characteristics of the scintillation light emission in pure LAr and the value of the rate constant k of the quenching process associated to the presence of N_2 contaminant [Sect.5.1].

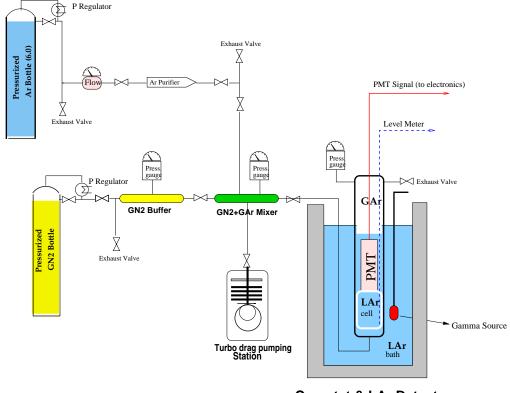
3 Detector and experimental method

A detector has been designed and assembled at the INFN-LNGS external facility ("Hall di Montaggio") for this specific task. It consists of a cylindrical LAr cell in PTFE (h=12 cm, \varnothing =8.5 cm internal dimensions, wall thickness 0.5 cm) containing about 0.7 lt of LAr (about 1 kg of active mass), viewed by a single 2" photo-multiplier (Electron Tubes ETL D745UA) mounted on the top end of the cell.

The photo-multiplier (PMT) is manufactured to work at cryogenics temperatures (LAr); however, the glass window is not transparent to the 127 nm LAr scintillation light. Therefore, the PMT optical window is covered by a layer of TetraPhenyl-Butadiene (TPB), an efficient frequency down-converter with emission spectrum peaked at 430 nm and width of about 50 nm (FWHM), well matching the PMT sensitivity range. To enhance the light collection a reflector layer (also covered with TPB) surrounds the internal walls (side and bottom) of the PTFE cell.

The detector is housed in a stainless steel cylindrical chamber (h=85 cm, \varnothing =10 cm), closed at both ends by vacuum-tight ConFlat (CF100) flanges, Fig. 1 (schematic layout). The internal volume of the chamber is about 6.5 lt and it contains, after filling, a total amount of 3.0 lt of LAr (including the LAr cell active volume at its bottom). The chamber is immersed in a LAr bath of a stainless steel open dewar, to liquify and keep at stable temperature the LAr internal volume.

A transfer line for the Ar filling and for the injection of controlled amounts



Cryostat & LAr Detector

Fig. 1. Schematic layout of the experimental set-up for the N_2 contamination test.

of gaseous Nitrogen (GN₂) has been also assembled and connected to the LAr chamber via vacuum-tight pipes and feed-throughs. The line includes a two-stages system formed by a gaseous Nitrogen (GN₂) buffer and a mixer of gaseous Nitrogen and Argon, connected via UHV valves, Fig. 1. The internal volume of the buffer is precisely known and accurate pressure control allows the injection of controlled amounts of N₂ gas in the LAr volume (from 1 ppm to 300 ppm of N₂ per injection with $\leq 6\%$ precision).

Gaseous Argon (GAr) for filling, supplied by a 200 atm pressurized bottle, is the best grade 6.0 (99.9999%) commercial Argon, with impurity concentration below 1 ppm. In Tab.1 the purity specification for the GAr in use is reported [16].

The GAr from the 6.0-bottle is flushed during filling through a Hydrosorb/Oxysorb cartridge positioned along the GAr line for partial removal 3 of O_2 and H_2O . The GN_2 for the controlled contamination is 5.5 GN_2 type (99.9995%), also supplied by a pressurized bottle.

In Fig. 2 a picture taken during the test is shown.

After vacuum pumping of the whole system (detector chamber and transfer

 $^{^3}$ No recirculation system is implemented for further O_2 purification. Initial content of residual N_2 in the 6.0 Ar is not filtered out. Additional contaminations from material out-gassing inside the chamber cannot be excluded.

Table 1
Argon 6.0 (from Rivoira Supplier) purity specifications: maximum impurity levels.

Oxygen (O_2)	≤ 0.2 ppm
Nitrogen (N_2)	$\leq 0.5~\mathrm{ppm}$
Water (H_2O)	$0.5~\mathrm{ppm}$
Hydrogen (H_2)	0.1 ppm
Total Hydrocarbons (THC)	$0.05~\mathrm{ppm}$
Carbon Dioxide (CO_2)	$0.05~\mathrm{ppm}$
Carbon Monoxide (CO)	$0.05~\mathrm{ppm}$

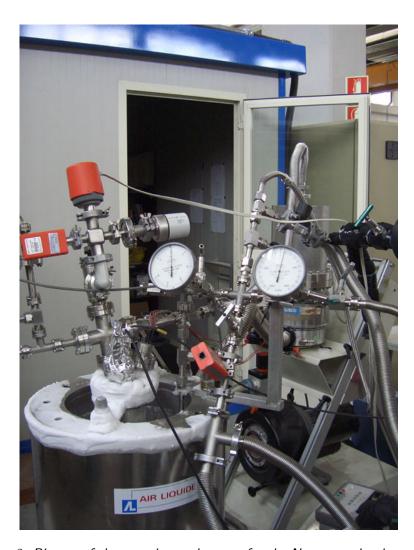


Fig. 2. Picture of the experimental set-up for the N_2 contamination test.

line) up to a residual vacuum of few 10^{-5} mbar, the external bath is filled with commercial LAr, the GAr line is opened and the purified Ar gas feeding the detector chamber is liquified inside it up to reaching a given level, precisely monitored by a level meter, corresponding to $3.0 \pm 1.5\%$ lt of LAr.

At LAr working conditions of T=86.5 K, P=930 mbar (ρ_{Ar} =1.40 g/cm³), contamination with GN₂ is performed by a controlled procedure: (1) GN₂ corresponding to a given fraction of the LAr molar quantity ⁴ in the chamber is first transferred into the GN₂ buffer volume (at controlled temperature and pressure), (2) the valve to the GN₂+GAr mixer is opened, (3) GN₂ is injected into the chamber by repeated flushes of purified GAr through the mixer. Ar and N₂ form almost ideal vapor/liquid mixtures ⁵. In the liquid phase they are completely miscible and almost ideal liquid solution. The vapor pressure in the top part of the chamber (about 3.5 lt above the liquid level) is different for Argon and Nitrogen and can be determined by the Antoine equation assuming the gas at the same temperature of the liquid Ar. Nitrogen results to be more volatile than Argon by a factor of 3; therefore the Nitrogen concentration in gas phase is higher than in the liquid [17]. However, since the gas/liquid volume ratio for Ar is very large (e.g. 835 at STP), the fraction of N₂ injected in the chamber which is diluted in the LAr volume is very close to one.

3.1 Data taking

The PMT read–out system is structured as an oscilloscope channel: in fact, the PMT+read-out chain can be represented by the simplified scheme of Fig. 3, where at the first stage the PMT signal is duplicated by a linear Fan-out, one output being used as input for a Constant Fraction Discriminator (CFD) providing a trigger signal when a defined threshold is passed, and the second output is directly recorded by a fast Waveform Recorder (Acqiris, DP235 Dual-Channel PCI Digitizer Card). At each trigger from the CFD the signal waveform is recorded with sampling time of 1 ns over a full record length of 10 μ s. A LabView application has been developed for the data acquisition and storage. In Fig. 4 a typical waveform recorded during the N₂ test is shown.

After the filling procedure with purified 6.0 Argon was completed, the N_2 experimental test started on February 2007 and lasted about one month. The test was performed by adding progressively controlled amounts of N_2 and exposing the LAr cell to γ -sources after each contamination. Data samples (10⁵ waveforms per run, trigger rate ~ 0.5 kHz) have been recorded in two

 $[\]overline{^4}$ The concentration units in use here are ppm, parts per million (atomic, often indicated as ppma). This indicates the ratio between the number of interesting elements (in our case N_2 molecules) to ordinary elements (Ar atoms).

⁵ The dilution process is rather slow. However, the speed is substantially increased by the repeated flushing of GN_2+GAr mixture through the liquid.

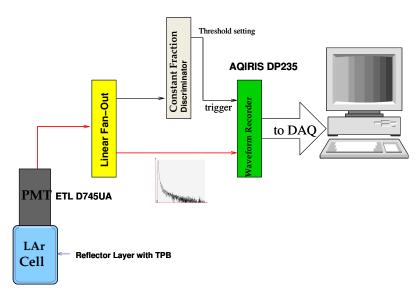


Fig. 3. PMT Electronics read-out chain.

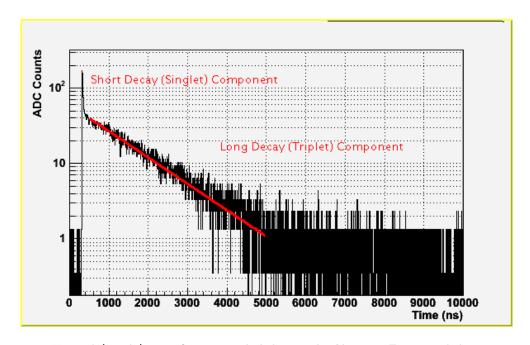


Fig. 4. Typical (single) waveform recorded during the N_2 test. Event with large energy deposition from cosmic muon (mip) crossing the LAr cell.

independent runs per N_2 concentration, with $^{137}\mathrm{Cs}$ and $^{60}\mathrm{Co}$ γ -ray sources. After the first run at 0 ppm (i.e. no additional N_2 injection to the initial purified 6.0 Argon), the contamination levels were set at $[N_2]=1$ ppm, 2 ppm, 7 ppm, 12 ppm, 20 ppm, 40 ppm, 60 ppm, 100 ppm, 500 ppm, 1000 ppm and 3000 ppm.

Single photo-electron measurements were also acquired before and after each γ -source run, to provide calibration data (see App. A) useful at various stages of the analysis performed.

4 Signal Amplitude Analysis

Gamma rays from the 60 Co source (1.27 and 1.33 MeV) induce mainly single Compton interactions in the LAr cell active volume (Compton edge $E_C \simeq 1$ MeV), with final state electron in the mip range. Scintillation light is emitted and, after down-conversion and reflections at the active volume boundaries, is collected at the PMT photo-cathode; finally the signal waveform is recorded. By single waveform integration, the absolute (individual) signal amplitude is

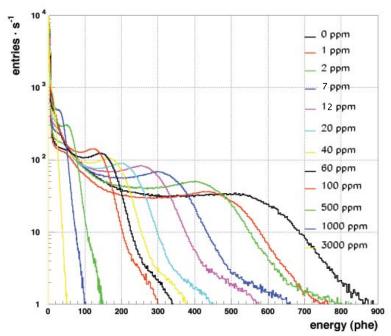


Fig. 5. ⁶⁰Co source Compton spectra from each [N₂] run.

obtained (in units of photo-electrons, eventually proportional to the electron energy deposited in the LAr cell).

Pulse amplitude spectra have been thus obtained for each run at different $[N_2]$ value, as reported in Fig. 5 with the 60 Co source. Manifestly, the spectra are progressively down-scaled at increasing N_2 concentration as due to the quenching process of reaction (5).

A dedicated fitting procedure has been developed to determine the scale factor Q_F (the overall Quenching Factor defined in Eq.11) acting on the absolute signal amplitude ⁶ as a function of the N_2 contamination. As an example, in Fig. 6 the result from the fitting procedure for the $[N_2]=20$ ppm run is reported $(Q_F([20 \text{ ppm}])=0.405$: i.e. only $\sim 40\%$ of Ar_2^* emit a photon, while the remaining $\sim 60\%$ undergo non-radiative N_2 collisions).

The Q_F values and associated errors obtained from the analysis of the whole

 $[\]overline{^6}$ Individual entry values of the uncontaminated (0 ppm) spectrum are down-scaled by Q_F , free parameter, to best fit the experimental $[N_2]$ contaminated spectra.

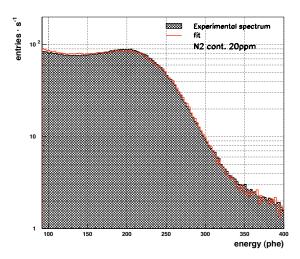


Fig. 6. ⁶⁰Co source for the $[N_2]$ =20 ppm run (dashed histogram). Superimposed the result of the fitting procedure (red histogram).

set of 60 Co runs at the various N_2 contaminations are given in Fig.7. After a fast drop up to ~ 100 ppm $(Q_F([100 \text{ ppm}])=0.25)$ a slower decrease is observed at higher N_2 concentrations (notice the *log-log scale* in use). The light contribution from the slow component is expected to be vanished above $[N_2] \geq 100$ ppm, and the fast component starts to be progressively affected by the quenching process because of the very high Nitrogen concentration.

The errors on the plot are from the fit (statistical) and are appreciable at the highest concentrations.

5 Signal Shape Analysis

As reported in Sec.3, data were collected by full signal waveform recording (1 ns sampling time over a time window of 10 μ s) for each trigger event. This allows a detailed reconstruction of the signal shape, in particular of the individual components (relative amplitude and decay time) of the scintillation light following an ionization event in LAr.

To this purpose the recorded waveforms (wfm) have been appropriately treated ⁷ and averaged ($\overline{V}(t)$ for each [N₂] run, 10⁵ events) and the averaged waveform amplitude (area) normalized to the corresponding Q_F value.

⁷ Individual waveform (wfm) are off-line processed: (1) 0.5 μ s of baseline presamples are used for baseline determination and subtraction, (2) cuts on wfm amplitude (integral) are applied to remove PMT saturations and low energy events, (3) wfm are time equalized at the same peak position and finally (4) selected wfm's are summed together.

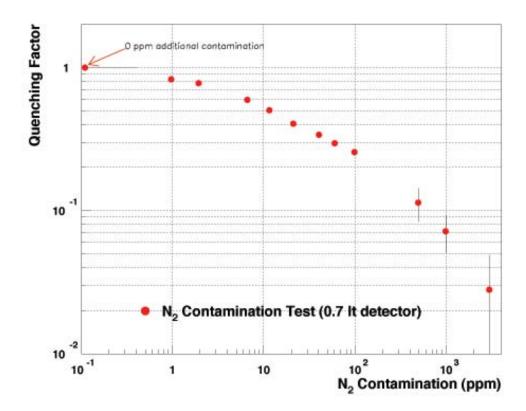


Fig. 7. Quenching factor Q_F as a function of the $[N_2]$ contamination (60 Co runs). The value corresponds to the fraction of Ar_2^* surviving N_2 quenching and producing VUV photons.

The averaged signal can be symbolically expressed as:

$$\overline{V}(t) = S(t) \otimes \mathcal{R}(t)$$
 (12)

where $\mathcal{R}(t)$ is the Response Function of the read-out system (PMT+cable+wfm digitizer), i.e. the electronic impulse response defined by the output signal obtained from an input single photo-electron pulse (usually indicated as SER, Single Electron Response).

The true light signal shape S(t) can be obtained from the recorded waveform $\overline{V}(t)$ by a standard deconvolution procedure [13], using the SER function $\mathcal{R}(t)$ experimentally obtained (as explained in App. A). In Fig.8 three averaged signals (at $[N_2]=0$ ppm, 12 ppm and 40 ppm) after SER deconvolution are shown.

The signal function $\ell'(t)$ (see Eq.7) convoluted with a gaussian function has been used to fit the signal shape S(t) from each $[N_2]$ run. The gaussian spread (free parameter $\sigma_G \simeq 4$ ns from fit) takes into account the time resolution of the light collection at the detector sensitive area, due to transit time of the PMT, wave length shifting time, time spread due light propagation including multiple reflections at the volume boundaries, etc.

Attempts to perform a fit with a signal function characterized by two time components (singlet and triplet) failed, as evident from Fig.9. This led to use a three components model (S, T and I, *Intermediate*), nicely fitting the present

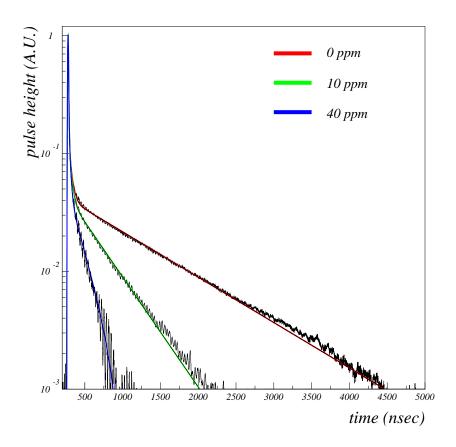


Fig. 8. Signal shape at 0 ppm, 12 ppm and 40 ppm of N_2 contamination, superimposed fits with three components.

data (color lines in Fig.8). The presence of an *intermediate* component is thus confirmed, in agreement with some earlier indications [11]. Dedicated studies on the origin of it are under way. Based on preliminary considerations, reported in App. B, possible instrumental effects appear as the main contribution to this component.

In Fig.10 the quenched amplitudes of the individual components A'_j from the fit are shown as function of N_2 contamination. The total amplitude (sum of the A'_j amplitudes at given $[N_2]$, black stars in Fig.10) is constrained to the corresponding Q_F value (Fig.7), as due to the average waveform normalization. The decay times τ'_j of the individual components from the fitting procedure of the signal shapes at different N_2 concentrations are reported in Fig.11.

5.1 Characteristics of LAr scintillation light and of N_2 quenching process

The quenched amplitudes and the decreased lifetimes of the scintillation light in N₂-contaminated LAr are correlated variables depending on the intrinsic

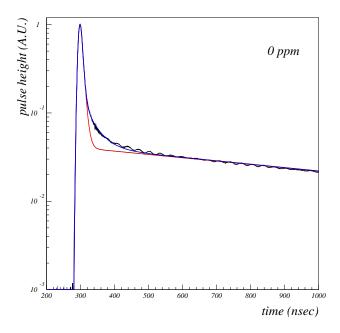


Fig. 9. Signal shape at 0 ppm contamination (black histogram) with comparison of two-components (red line) vs. three-component (blue line) model.

LAr scintillation characteristics and on the rate constant of the N_2 quenching process, as reported in Sec.2.1.

An overall fit of the data obtained from the signal shape analysis reported in Fig.10 and Fig.11 has been performed using Eq.9 and Eq.8 as fitting functions, respectively. The values of eight free parameters have been thus obtained by χ^2 -minimization: the LAr scintillation characteristics (intrinsic lifetimes τ_j and relative amplitudes A_j , j = S, T, I for mip electronic recoils), the rate constant k of the quenching process due to N_2 contaminant, and the initial Nitrogen concentration $[N_2]_{in}$. This last corresponds to the N_2 content in the 6.0 grade Argon used for the initial filling. Its value is obtained by using an additive free parameter in the fitting functions: $[N_2] \rightarrow [N_2] + [N_2]_{in}$.

A contamination of Oxygen at the level of $[O_2]$ =0.06 ppm has been taken into account into the fit, as determined by the analysis reported in [3] from the parallel test of Oxygen contamination effects.

The values of the parameters obtained from the overall fit are reported in Tab.2.

Table 2
Results from the overall fit: characteristics of scintillation radiation in LAr from mip electronic recoil and of quenching process from Nitrogen contamination. Quoted errors are statistical from the fit.

Lifetimes	Amplitudes	N_2 rate constant	Initial N ₂ concentration
$\tau_S = 4.9 \pm 0.2 \text{ ns}$	$A_S = 18.8\%$		
$\tau_I = 34 \pm 3 \text{ ns}$	$A_I = 7.4\%$	$k = 0.11 \pm 0.01 \text{ ppm}^{-1} \mu s^{-1}$	$[N_2]_{in} = 0.40 \pm 0.20 \text{ ppm}$
$\tau_T = 1260 \pm 10 \text{ ns}$	$A_T = 73.8\%$		

The rate constant 8 of the N_2 quenching process (k value) is in agreement with an earlier measurement found in literature [14].

The initial N_2 concentration is within the expected range from the specifications of the 6.0 grade Ar used for filling.

The relative amplitude of the intermediate component is small $(A_I = 7.4\%)$ compared to the amplitude of the other two components. It is also worth noticing that the amplitude ratio between $A_S + A_I$ and A_T is 0.35 in agreement with available reference data for light ionizing particles [11].

In all components the fit is found to deviate from the data at the highest N_2 concentrations, as possibly due to incomplete absorption of impurities into the liquid. This trend might be explained by a *saturation* effect of the solute (Nitrogen) in the LAr solvent. Similar indications have been reported by other groups [14] and also in our test with O_2 contamination [3]. The measurements were performed on relatively short time scale (hours to day per contamination). This prevents to draw any definite conclusion on long term effects.

A different parameterization depending on the N_2 concentration under saturation hypothesis has been thus used for the fitting function of the lifetimes (each component):

$$\frac{1}{\tau_i'} = \frac{1}{\tau_j} + k \beta \left(1 - e^{-\frac{[N_2]}{\beta}} \right) \tag{13}$$

and of the amplitudes:

$$A'_{j} = \frac{A_{j}}{1 + \tau_{j} k \beta (1 - e^{-\frac{[N_{2}]}{\beta}})}$$
(14)

where the β parameter represents the concentration scale where saturation becomes effective ($\beta = 530$ ppm from fit). At first order approximation both formulas reduce to the original fitting functions (Eq.8 and Eq.9).

The curves in Fig.10 and Fig.11 (full line in corresponding color) show the result of the fit with the modified parameterization. This model improves the scaling with N_2 concentration and decreases the overall χ^2 , without changing at all the results reported in Tab.2.

⁸ The rate constant of the two-body (N₂-Ar₂*) process is assumed to be the same for both the singlet ${}^{1}\Sigma_{u}$ and the triplet ${}^{3}\Sigma_{u}$ Ar excimer states. A fit has been tried with two rate constants as free parameters. This indicates that in case of N₂- ${}^{1}\Sigma_{u}$ coupling, a slightly higher constant rate is preferred.

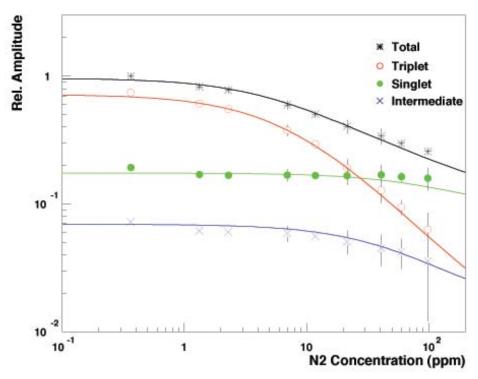


Fig. 10. Effective amplitudes (mip electronic recoil) of the individual components $(A'_j, j = S, T, I)$ in corresponding color an symbol) as function of N_2 concentration. The total amplitude (sum of the A'_j amplitudes at given $[N_2]$) is constrained to the corresponding Q_F value (black stars) obtained in Sec.4. Lines in corresponding color are from the fit with saturation model (Eq.14).

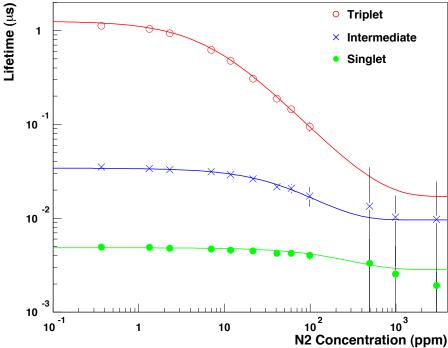


Fig. 11. Effective decay times of the individual components $(\tau'_j, j = S, T, I)$ in corresponding color an symbol) as function of N_2 concentration. For $[N_2] \ge 500$ ppm, $\tau'_T \to 0$ (not shown). Lines in corresponding color are from the fit with saturation model (Eq.13).

The errors associated to the results in Tab.2 are from the fit (statistical). These take into account the statistical errors from the fit of the waveforms at different N_2 contaminations and also the error associated to the injected amounts of Nitrogen in the contamination procedure.

The goodness of the overall fit (saturation model) is satisfactory (C.L. $\simeq 80\%$). Systematic uncertainty may come from different sources: instability of the signal read-out system, bias due to residual concentration of unknown types of impurities in LAr leading to light absorption or further Ar-excimer quenching, possible uncertainty introduced by the procedures of data analysis. Only this last issue has been investigated in some details.

The off-line analysis of the light signal shape (obtained from the PMT waveform recording at the different N_2 contaminations) is based on the deconvolution procedure of the SER function (see App.A). Systematic uncertainty associated to this procedure is relevant mainly to the long-lived component determination and has been estimated to be around 4%: $[\tau_T = 1260 \pm 10 \ (stat) \pm 55 \ (syst) \ ns]$. Comparisons with alternative fit procedures confirm this evaluation, as reported in details in App.C.

6 Conclusions

New generation detectors based on LAr as active medium for Dark Matter direct search and for neutrino physics exploit both scintillation light and free electron charge from ionization events.

Residual content of electro-negative impurities in LAr (like Oxygen molecules) significantly reduces the amount of charge and light by attachment and quenching processes respectively. This led to the development of dedicated O₂ purification systems, routinely employed in the experimental set-up's.

On the other hand, non electro-negative contaminants like Nitrogen can also be found at appreciable concentration level even in best grade commercial Argon. Contamination at ppm level of N_2 leads to a substantial reduction of the scintillation light intensity, due to the quenching process in two-body collision of N_2 impurities with Ar_2^* excimer states otherwise decaying with VUV light emission. (No appreciable effects from N_2 contamination on the free electron charge can be expected, due to the low electron affinity of the N_2 molecule).

A dedicated test has been performed by means of a controlled N_2 contamination procedure. Measurements have been done by mip particle excitation (Compton electrons from γ -sources in the MeV range).

The effect on the scintillation light collection has been measured over a wide range of N_2 concentration, spanning from $\sim 10^{-1}$ ppm up to $\sim 10^3$ ppm, though a *saturation* effect of the solute (Nitrogen) in the LAr solvent has been presumably found at the highest contaminations.

The rate constant of the light quenching process induced by Nitrogen in LAr has been found to be $k(N_2)=0.11~\mu s^{-1} ppm^{-1}$ (in agreement with some early measurements reported in literature). This implies that, for example, at ~ 1 ppm level a $\sim 20\%$ reduction of the scintillation light is experienced due to the N_2 quenching process. The quenching factor affecting the light yield has been determined over the whole explored range of N_2 concentrations.

The experimental test also allowed to extract the main characteristics of the scintillation light emission in pure LAr, thanks to the direct PMT signal acquisition by the fast Waveform Recorder in use with the implemented DAQ system.

The light signal shape is well represented by the superposition of three components with exponential decay. The fast and the slow components, with decay time constants τ_S =4.9 ns and τ_T =1260 ns, are recognized to be associated to the decay of the singlet $^1\Sigma_u$ and the triplet $^3\Sigma_u$ excimer states of Ar_2^* . An intermediate component is also found to be present (as sometime reported in literature), whose origin could presumably be ascribed to PMT instrumental effects.

The main effect of residual N_2 in LAr is of reducing the slow component lifetime at increasing concentration, and consequently of varying the ratio of the relative amplitudes to higher values. This makes the slow signals from γ/e background less distinguishable in Pulse Shape vs the fast Ar-recoils signals possibly induced by WIMP interactions, in Dark Matter search experiments. Therefore, the implementation of dedicated methods for removal of the residual N_2 content results to be recommended with LAr based detectors, or at least the use of best grade commercial Argon with reduced Nitrogen contamination (below ~ 0.5 ppm) appears as definitively necessary.

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A Single Electron Response

Single photo-electron pulses (SER data) from thermionic dark counts have been routinely acquired during the test, before and after each source run at the different N_2 concentrations.

Pulse amplitude spectra from SER waveform integration were used for

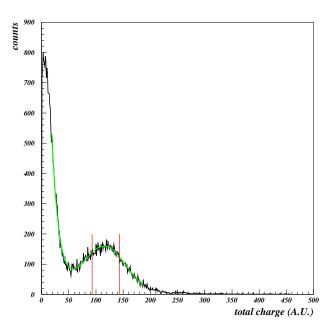


Fig. A.1. SER spectrum of the PMT in use and fit superimposed. The interval around the mean value for single photo-electron waveform selection is also indicated.

calibration purposes. Typical SER spectrum of the PMT in use is reported in Fig.A.1: the Gaussian distribution at higher ADC values corresponds to the genuine single photo-electron mean amplitude and spread. The mean value (gaussian fit) gives the calibration constant, from ADC units into photoelectrons units, useful for Compton spectra analysis (Sec.4) and checking of the detector stability in time.

The single photo-electron pulses collected by full signal waveform recording (1 Gsample/s over 10 μ s time window) were used for the determination of the read-out electronics *Response Function*: the impulse response $\mathcal{R}(t)$ is obtained by processing and averaging the SER waveforms with amplitude in a defined interval around the mean value of the corresponding single photo-electron pulse amplitude spectrum.

A typical SER waveform averaged on a large sample of recorded signals is shown in Fig.A.2. This is characterized by two main features: (1) the genuine single photo-electron shape from the PMT with FWHM $\simeq 10$ ns and a decay tail extending up to about 150 ns, and underneath (2) a very long, slowly decreasing component up to $\sim 3\mu$ s (visible in the up-right, log scale panel of

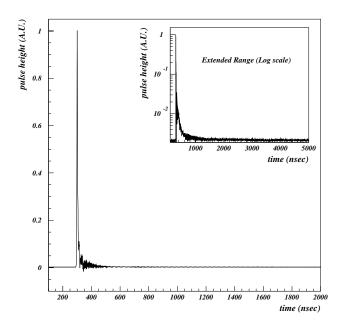


Fig. A.2. Example of SER averaged waveform $\mathcal{R}(t)$.

Fig.A.2) followed by a flat baseline.

The flat baseline is due to pile-up (in the average wfm) from the PMT dark counts randomly distributed in the recorded time window, while the origin of the long-decay component is not fully understood. It may be ascribed to additional pile-up of after-pulses from ionization of residual gases in the PMT (normally reported in PMT test reports [18]). Typically, these type of after-pulses are delayed in a $\mathcal{O}(0.5$ - 5) μ s interval after the main signal from which they originate, in the ratio of $\leq 5\%$ to the number of true pulses.

The SER function $\mathcal{R}(t)$ (including the pile-up components) determined for each source run is used in the deconvolution procedure to obtain the corresponding true light signal shape S(t) (Sec.5).

The deconvolution of the response function $\mathcal{R}(t)$ from the experimental waveform $\overline{V}(t)$ is performed numerically in the frequency domain. The discrete Fourier transforms \tilde{v}_i and \tilde{r}_i of \overline{V}_i and \mathcal{R}_i respectively are first evaluated by the Fast Fourier Transform (FFT) algorithm [13]. According to the convolution theorem, the true light signal S_i is hence obtained by taking (with the same FFT algorithm) the inverse Fourier transform of \tilde{s}_i (= \tilde{v}_i/\tilde{r}_i).

B The intermediate component: preliminary considerations

As described in Sec.5, evidence for an intermediate component in the scintillation signal shape has been found at all the N_2 contaminations, including the "0 ppm run" with lifetime $\tau_I \simeq 34$ ns and relative amplitude

- $A_I \simeq 7.4\%$. The origin of this component is controversial: it could be ascribed to (1) instrumental effects (in the PMT), (2) presence of residual N₂ concentration in LAr, (3) physical (unknown) properties of Argon. Explanations in terms of:
- Intrinsic physical properties of Argon yet unknown or only marginally investigated cannot be excluded. For example, from the delayed decay of vibrationally hot Ar_2^* molecules surviving vibrational relaxation [8]. This decay, seen in gas at low pressure, was never investigated in liquid phase.
- Light emission from Nitrogen trace in LAr is possible though rather unlike in the present experimental conditions: (a) scintillation emission from excited Nitrogen molecular states $N_2^*(C^3\Pi_u)$ formed by energy transfer interaction with Ar* atoms, as seen in gaseous Ar-N₂ mixtures, should be negligible since in liquid phase the precursor Ar* atoms are quickly self-trapped ⁹, leading to Ar₂* formation (moreover no increasing A_I amplitude is found at increasing N₂ concentrations in Ar-N₂ studies [this paper] and [14]), however (b) a series of mechanisms leading to the formation of excited Nitrogen atomic states from Ar₂* energy transfer has been recently identified and detected in Ar-N₂(few %) gas mixtures [19]. The N* radiative decay (149 nm and 174 nm) may thus populate the intermediate component amplitude. The size of this effect, presumably small in liquid mixtures with very low N₂ concentrations, as in the present study, is of interest for further studies.
- Instrumental effects to explain the onset of the Intermediate component is our preferred option. Such effects could be ascribed to the use of PMT and enhanced by particular experimental conditions. For example, the so called anode glow in the PMT [18] can be considered: it has long been known that electrons in PMTs induce light emission in the last cascades of dynode systems. Photons may take the way back to the photo-cathode and so give rise to afterpulses, typically 20 to 50 ns after the true pulse from which they originate (and up to 120 ns in large PMTs). This effect has been well characterized in various studies, indicating that the total amount is of the order of 1%.

Contributions from *late-pulses* due to photo-electron backscattering effect on the first dynodes can also be taken into account. Late-pulses are part of the main pulses but delayed by 5 to 80 ns, in the ratio of $\leq 5\%$ to the number of true pulses [18].

Both contributions may increase in high photo-electron density conditions, as in our case induced by the LAr fast decay component in the first few ns of scintillation emission.

All these are preliminary considerations. Only dedicated studies (outside the limits of this report) may possibly indicate the true origin of this component.

The Ar* self-trapping reaction rate is $\sim 10^{12} \ \rm s^{-1}$ [14], while the collision frequency of the energy transfer process is $\sim 10^6 \ \rm s^{-1}$ at [N₂]=0.5 ppm (e.g. like in our "0 ppm run").

C Systematic error in the slow component lifetime determination

After completion of this report and while proof reading, an exhaustive paper on Scintillation time dependence and pulse shape discrimination from Lippincott et al. [20] appeared in publication. In this paper a value for the slow component lifetime of τ_T =(1463 ± 5_{stat} ± 50_{syst}) ns is quoted. This value is only marginally compatible with the one we found here of τ_T =(1260 ± 10_{stat} ± 55_{syst}) ns.

We ascribe this discrepancy to the different strategies adopted in the analysis of the LAr scintillation waveforms and not to physical or instrumental effects. In fact, by applying the procedure described in [20] to fit the tail of the average waveform $\overline{V}(t)$ obtained from the "0 ppm run" data sample we find a τ_T value for pure LAr in full agreement with the one reported in [20].

In more details, the value of the slow component lifetime we obtain with the fit model from [20] (characterized by an additional flat baseline component)

$$\overline{V}(t) = A \exp(-t/\tau_T) + B \tag{C.1}$$

is $\tau_T = (1306 \pm 5)$ ns. Taking into account that the Argon we used to fill the cell ("0 ppm run") had indeed some residual contamination of Nitrogen and Oxygen (400 ppb of N₂ and 60 ppb of O₂) and having determined the values of the quenching rate constants k for N₂ [this report] and k' for O₂ [3] respectively, we can extrapolate the τ_T value for pure LAr, that results to be $\tau_T = (1453 \pm 10)$ ns.

In Fig.C.1 we show the average waveform $\overline{V}(t)$ with the fit (C.1)

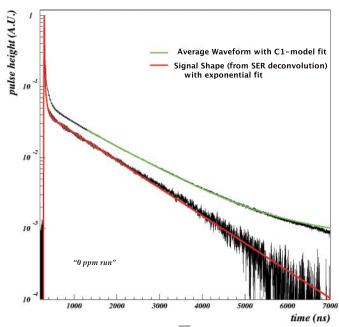


Fig. C.1. Fit comparison: average waveform $\overline{V}(t)$, as built by summing up the waveforms recorded in the "0 ppm run", with C.1-model fit (green line) and signal shape S(t) (standard deconvolution procedure) with exponential fit (red line).

superimposed. For comparison the light signal shape S(t) as obtained from $\overline{V}(t)$ with our standard procedure based on the SER function $\mathcal{R}(t)$ deconvolution is also shown. The difference in the slope of the long-component is quite evident. In our data this is due to the presence of an "extralong tail" exponentially decaying with lifetime of $\sim 3~\mu s$ (rather than the simple flat baseline component), also present in the SER waveform (Fig.A.2). This possibly may fake a longer triplet state decay time in the average non-deconvoluted waveform. Thanks to the deconvolution procedure this component is washed out.

It is worth noting that when alternatively a fit of V(t) with an extra-long decay component is performed $(B \to B \ exp(-t/\tau_B))$ in Eq.C.1, with $\tau_B \simeq 3 \ \mu s$) the triplet component lifetime is determined within 2% spread around the value obtained from our standard procedure (SER deconvolution).

To further explore possible sources of systematic uncertainty due to the offline procedures in our data analysis, we also tested the classic method of *Coincidence Single Photo-electron Counting* [21] for another determination of the long component lifetime.

The Coincidence Single Photo-electron Counting technique has been largely exploited in many reports on scintillation lifetime measurements (for LAr see [11],[12],[14]). Data recorded with our set-up (one single PMT with direct signal acquisition by fast Waveform Recorder) allow to perform an artificial single photo-electron counting experiment, based on off-line data treatment. Starting at 0.5 μ s after the onset of a triggered signal, a single photo-electron identifying algorithm was run through the recorded waveform V(t) and for each photo-electron pulse (defined by appropriate cuts) found at time t_i a single bin with the value of 1 was substituted.

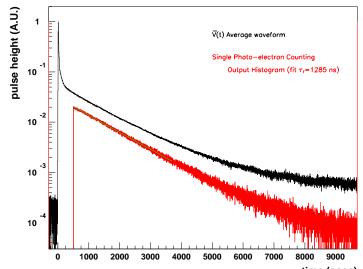


Fig. C.2. Result of the artificial single photo-electron counting method for τ_T determination (red histogram) with the "0 ppm run" data sample. The average waveform $\overline{V}(t)$ (black histogram) is also shown for comparison.

As an example with the "0 ppm run" data sample, the resulting histogram, which is per se cleaned from noise contributions, provides a τ_T determination that is in very good agreement (within $\sim 2\%$) with the result of our standard procedure based on the SER function deconvolution, as shown in Fig.C.2.